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**Elementarzellen und Raumgruppen der Peptidderivate Carbobenzoxy-L-leucyl-L-tyrosyl-L-leucinmethylester und Carbobenzoxyglycyl-L-alaninäthylester.\*** Von W. L. HAAS†, *Deutsches Wollforschungsinstitut an der Technischen Hochschule, Aachen, Germany*

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Gillesen (1961) synthetisierte das Peptid L-Leucyl-L-tyrosyl-L-leucin, dessen Derivat Carbobenzoxy-L-leucyl-L-tyrosyl-L-leucinmethylester aus Essigester in verhältnismässig gut ausgebildeten gedrunenen hexagonalen Pyramiden kristallisierte. Ausgebildet waren ausschliesslich (001)- und (10 $\bar{1}$ 0)-Flächen. Eine Drehkristallaufnahme um die *c*-Achse mit vanadiumgefilterter Chromstrahlung und Präzessionsaufnahmen nach Buerger mit den Einstrahlungsrichtungen parallel zur *a*-Achse sowie parallel zur Winkelhalbierenden zweier Nebenachsen mit nickelgefilterter Kupferstrahlung ergaben eine hexagonale Elementarzelle mit den Translationsperioden

$$a_0 = b_0 = 7,22 \pm 0,02; c_0 = 36,31 \pm 0,1 \text{ \AA}$$

und den Winkeln  $\alpha = \beta = 90^\circ$ ;  $\gamma = 120^\circ$ .

Daraus wird unter Annahme von 6 Molekülen in der Elementarzelle die Dichte zu 1,12 g.cm.<sup>-3</sup> berechnet. Gefunden wurde nach der Verdrängungsmethode 1,11 g.cm.<sup>-3</sup>. Die Laue-Symmetriegruppe  $\bar{6}/mm$  und die ausschliesslich beobachteten 00*l*-Reflexe mit  $l = 3n$  lassen die Wahl zwischen den Raumgruppen  $P6_22$  und  $P6_422$ . Im reziproken Gitter treten parallel zur *c*-Achse zwischen den Schichtlinien schwache und diffuse Reflexserien auf, deren Bedeutung und Herkunft unklar ist.

\* 25. Mitt. über Peptide; 24. Mitt., vgl. H. Zahn & M. Heinz, *Liebigs Annalen*, im Druck.

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Im Zusammenhang mit Arbeiten zur Synthese von Peptiden mit Sequenzen des Seidenfibroins fiel das Peptidderivat Carbobenzoxyglycyl-L-alaninäthylester in Form nadeliger Einkristalle an. Eine Drehkristallaufnahme sowie Weissenbergaufnahmen um die Nadelachse (*c*-Achse) der rhombischen Prismen und Präzessionsaufnahmen nach Buerger mit der Einstrahlungsrichtung parallel zu den Flächendiagonalen der Prismenendfläche (*a*- und *b*-Achse) ergaben eine orthorhombische Elementarzelle mit folgenden Dimensionen:

$$a_0 = 21,17 \pm 0,05; b_0 = 9,78 \pm 0,03; c_0 = 16,10 \pm 0,04 \text{ \AA}$$

und den Winkeln  $\alpha = \beta = \gamma = 90^\circ$ .

Es waren praktisch ausschliesslich (001)- und (110)-Flächen ausgebildet. Allgemeine systematische Auslöschungen traten nicht auf. Die fehlenden Reflexe  $h00$  mit  $h = 2n + 1$ ,  $0k0$  mit  $k = 2n + 1$  und  $00l$  mit  $l = 2n + 1$  lassen auf die Raumgruppe  $P2_12_12_1$  schliessen.

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**The atomic position parameter in alpha uranium-room temperature and above.** By MELVIN H. MUELLER, RICHARD L. HITTERMAN, and HAROLD W. KNOTT, *Argonne National Laboratory, Argonne, Illinois, U.S.A.*

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Alpha uranium, which is orthorhombic, space group  $D_{2h}^{17}$ - $Cmcm$ , with 4 U per unit cell at  $0, y, \frac{1}{4}$ ;  $0, \bar{y}, \frac{3}{4}$ ;  $\frac{1}{2}, \frac{1}{2} + y, \frac{1}{4}$ ;  $\frac{1}{2}, \frac{1}{2} - y, \frac{3}{4}$ ; can be described in terms of corrugated sheets parallel to the *b* face in which the *y* parameter is a measure of the degree of corrugation—the larger the value of *y* the greater the degree of corrugation. It becomes of interest to determine the change in *y* with temperature since an increase in *y* makes the orthorhombic uranium lattice more nearly hexagonal.

Previously reported and presently determined values of *y* are shown in Table 1. Most of the results are in fairly good agreement near room temperature, however, the value of *y* at elevated temperatures has not been very thoroughly investigated. Therefore this investigation was undertaken to determine especially values at the elevated temperatures.

For the space group concerned the structure factor, *F*, can be stated as  $F = 4(f_U \text{ or } b_U) \cos 2\pi hx \cos 2\pi ky \cos 2\pi lz$ , where  $f_U$  is the X-ray and  $b_U$  the neutron scattering

factors for uranium. Since there is only one positional parameter, *y*,  $F = 4(f_U \text{ or } b_U) \cos 2\pi ky$ , therefore the (0*k*0) reflections are particularly suitable for its determination. This technique was used recently by Sturcken & Post (1960) in their determination of the *y* positional parameter at 25 °C. Since their technique involved the determination of a minimum value for the agreement factor, *R*, for selecting the best value of *y* using a predetermined value of the temperature factor, *B*, it was decided at first to recheck the room temperature values of *y* and *B* using the Busing-Levy (1959) least-squares program.

X-ray data were first obtained at room temperature in a manner similar to that used by Sturcken & Post (1960) using a small single crystal approximately 3 mm. on an edge with a polished (0*k*0) face. These results together with a redetermination of *y* from the Sturcken & Post data using the Busing-Levy program are shown in Table 1.

The uranium crystal was then mounted in a high

Table 1.  $y$  Positional parameter for alpha uranium as reported by various investigators

Investigator	Material	Temperature	$y$
Jacob & Warren (1937)	Powder	R.T.	$0.105 \pm 0.005$
Konobeevsky <i>et al.</i> (1958)	Powder	20 °C.	$0.107 \pm 0.003$
Chebotarev (1961)	Powder	R.T.	0.105
Sturcken & Post (1960)	Single crystal	25 °C.	$0.1025 \pm 0.0003^*$
Cash <i>et al.</i> (1961)	Powder	R.T.	$0.102 \pm 0.002$
Present invest. (X-ray)	Single crystal	25 °C.	$0.1025 \pm 0.0005$
Present invest. (neutron)	Single crystal	25 °C.	$0.1024 \pm 0.0003$
Konobeevsky <i>et al.</i> (1958)	Powder	500 °C.	$0.115 \pm 0.003$
Chebotarev (1961)	Powder	640 °C.	0.112
Present invest. (neutron)	Single crystal	625 °C.	$0.1057 \pm 0.0006$

\* A value of  $0.1024 \pm 0.0005$  was obtained as redetermined with the same data using the least-squares program.

temperature furnace designed primarily for polycrystalline X-ray studies. The crystal was held in an open section in the center of a large silver plate by very light tantalum springs. Silver served as a good holder since it is a good thermal conductor and does not react with uranium. The plate and crystal with suitable thermocouples were held in the center of a resistance furnace. An ion getter pump attached to the top of the furnace eliminated all vacuum lines during runs and produced a vacuum of  $10^{-6}$  to  $10^{-7}$  mm. Hg.

Except for a few preliminary X-ray measurements, data were obtained by neutron diffraction since it had advantages over X-ray diffraction such as: larger volume sampling, smaller absorption by the surface irregularities and inappreciable surface oxide effect, all due to the greater penetration of neutrons; no drop off of neutron scattering amplitudes as a function of angle; and easier alignment of the sample due to larger instrument geometry.

The furnace with crystal was then carefully aligned

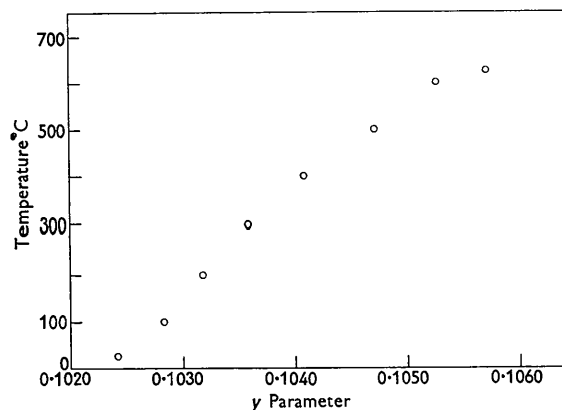


Fig. 1. Plot of  $y$  positional parameter for alpha uranium as a function of temperature.

on the neutron diffractometer. The correct  $\theta$  angle was established by means of an adjustable omega motion in the base of the furnace and the diffracted beam was centered in the counter by means of a shutter assembly which permitted observing the left, right, top and bottom quadrants separately. Integrated intensities were then obtained by step scanning  $2\theta$  using a moving crystal-moving counter technique.

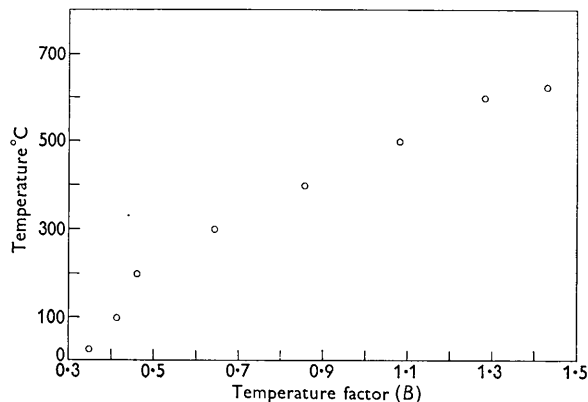


Fig. 2. Plot of temperature factor  $B$  for alpha uranium as a function of temperature.

Intensity data were obtained for the (040), (060), (080) and (0,10,0) for approximately 100 °C. intervals using a neutron wavelength of 0.98 Å. The (020) was partially masked by the edge of the furnace window.

Plots of  $y$  and  $B$  versus temperature are shown in Figs. 1 and 2 respectively, and both  $y$  and  $B$  appear to rise gradually with some flattening near 625 °C. Since the interdependence of these two variables was of interest, additional calculations were made on the 625 °C. data holding  $B$  constant at 1.1 and 0.8. The corresponding  $y$  values were 0.1057 and 0.1058 indicating little dependence. As shown in Table 1 the room temperature  $y$  values determined by X-ray and neutron diffraction are in excellent agreement. The results obtained with neutron diffraction at elevated temperatures, however, are considerably lower than the previously reported values.

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